

SUMMER – 19 EXAMINATION

Subject Name: Heat power Engineering

<u>Model Answer</u>

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Important Instructions to examiners:

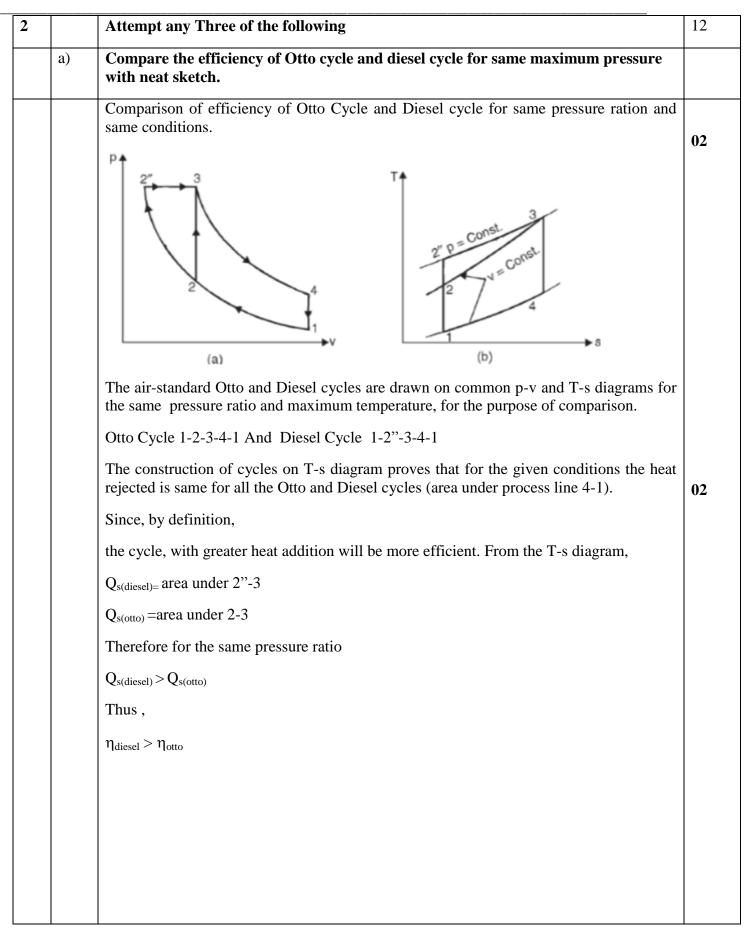
- 1) The answers should be examined by key words and not as word-to-word as given in the model answer scheme.
- 2) The model answer and the answer written by candidate may vary but the examiner may try to assess the understanding level of the candidate.
- 3) The language errors such as grammatical, spelling errors should not be given more Importance (Not applicable for subject English and Communication Skills.
- 4) While assessing figures, examiner may give credit for principal components indicated in the figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for any equivalent figure drawn.
- 5) Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answers and model answer.
- 6) In case of some questions credit may be given by judgement on part of examiner of relevant answer based on candidate's understanding.
- 7) For programming language papers, credit may be given to any other program based on equivalent concept.

Q. No	Sub Q. N.	Answer	Mark ing Sche me
1		Attempt any Five of the following.	10
	a	Draw P-V diagram of Otto cycle.	
		f(d) = C $Volume(V)$	02
	b	List the different types of thermodynamic system.	02
		1- Open system	
		2- Closed system	
		3- Isolated system Define calorific value of fuel and state its unit.	02
	c	"Calorific value" of fuel:	02
		It is defined as the amount of heat liberated during complete combustion of 1 kg of fuel. It is expressed in terms of KJ/kg.	01



d	State any four requirements of good fuel	02
	1. It should have high calorific value.	1/2
	2. An ideal fuel should have moderate ignition temperature.	mark for
	3. Its moisture content should be low.	each
	4. It should evolve large amount of heat when it burnt.	
	5. Moderate velocity of combustion.	
	6. Easy to transport	
	7. Low cost.	
e	Write the PVT relation for adiabatic process and give the meaning of each suffix used.	02
	$P_2/P_1 = (V_2/V_1)^{\gamma} = (T_2/T_1)^{\gamma/\gamma} -1$	
	γ =adiabatic index	
	P_1 , P_2 = pressure at inlet and outlet	
	V_1, V_2 = volume at inlet and outlet	
	T_1, T_2 = temperature at inlet and outlet	
f	Write any four application of compressed air in automobile industry.	02
	1. Operating tools in factories	
	 2. Operating drills and hammers in road building 3. Starting diesel engines 	1/2 mork
	4. Operating brakes on buses, trucks and trains	mark for
	5. Spray painting6. Excavating	each
	7. To clean the large workshops	
g	Enlist non-conventional energy sources.	02
	Solar power	
	Hydro-electric power	
	Wind power	
	Tidal power	
	Ocean wave power	
	Geothermal power	
	Ocean thermal power	
	Biomass, Bio-fuel etc.	





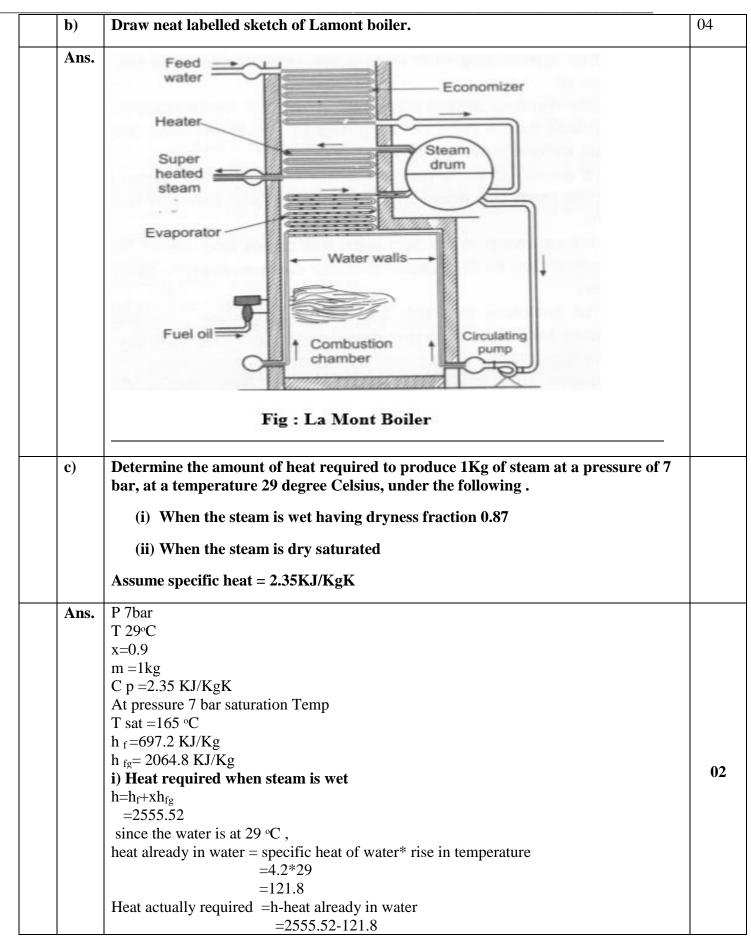


Ī	b)	Write the classification of air compressor.	04
-		COMPRESSORS	
		POSITIVE DISPLACEMENT DYNAMIC	
		RECIPROCATING ROTARY CENTRIFUGAL AXIAL	
		SINGLE-ACTING HELICAL-SCREW SCROLL	
		SINGLE-ACTING INCLOAL-SCREW	
		DOUBLE-ACTING LIQUID-RING SLIDING-WAVE	
F	c)	One kg of gas occupying 0.1m ³ at pressure of 14 bar is expanded at constant	
	()	pressure to 0.2m ³ . Determine an initial and final temperature of gas. Take	
		Cp=1.008KJ/KgK, Cv =0.72KJ/KgK.	4
-			
		$V_1=0.1m^3$ $V_2=0.2m^3$	
		$P_1 = P_2 = 14$ bar	
		$C_p=1.008 \text{ KJ/KgK}$	
		$C_v = 0.72 \text{ KJ/KgK}$	02
			02
		R=Cp-Cv	
		R=1.008-0.72 R=0.288KJ/KgK	
		Characteristic gas equation,	
		$P_1V_1 = mRT_1$	
		14*10^5*0.1=1*288*T1	02
		T ₁ =486.11K	02
		For constant pressure process,	
		V1/T1=V2/T2 0.1/486.11=0.2/T2	
		$T_2=972.22K$	
F	d)	Write factors used for governing the selection of cogeneration system and state the	04
	,	advantages of cogeneration.	
F		Factor governing the selection of cogeneration system:	
		1) electrical load matching	
		2) thermal load matching	
		3) Electrical load matching	
		4) Thermal load matching	
		5)Thermal to electrical energy ratio	02
		6) Quality of required thermal energy7) Fuel availability	
		8) Installation and available space	
		9) Operational costs	
			1



		 10) Pollution concern Advantages of Cogeneration: 1) Co-generation can meet both power & heat needs. 2) Less costly. 3) Very high efficiency. 4) Reduction in emission of pollutants due to reduced fuel consumption. 	02
3		Attempt any THREE of the following.	12
	a	Explain application of conduction and convection mode of heat transfer in automobile.	04
		 (Any suitable Answer should be given full marks) Applications of conduction- Fins provided on motor cycle engine Electric fuse cut off Electric heater Carbonisation of coal Melting of iron in blast furnace Fission reactions in nuclear fuel rods of nuclear reactors. Electric discharge machining in manufacturing Applications of convection- Forced Convection is used to cool down the headed plate . Forced Convection is used to cool down the laptop and super computer etc. Forced convection is used to cool down human body in summer season. Radiator - Puts warm air out at the top and draws in cooler air at the bottom. 	02







		=2433.72KJ/Kg	
		ii) When the stem is dry saturated $h_g=h_f+h_{fg}$ =2762 kj Heat actually required = 2762-121.8 = 2640.2 kj	
			02
	d)	Explain solar power plant and write its two advantages.	04
	Ans.	Construction: The basic components of solar power plant are also exactly identical to thermal power plant except the boiler is replaced by solar collector. The arrangement of component is as shown in figure. The energy from solar radiation is collected and utilized to generate steam to run the turbine. Solar reference of the turbine for the turbine for the turbine of the turbi	02
4		Attempt any THREE of the following	12
	a)	Write the importance and impact of energy conservation on environment and economy.	04
		 (Any suitable Answer should be given full marks) Impact of energy conservation on Economy-Reduced dependence on non-renewable sources of energy: Based on current known reserves and consumption of these fuels. It protects the economy and consumers from possible price fluctuations and from energy service disruptions due to natural disasters or other 	02



	$C + O_2 \longrightarrow CO_2$ i.e. $12 + (16 \times 2) = 12 + 16 \times 2$ i.e. $12 + 32 = 44$	
c)	Explain combustion chemistry of carbon, methane and hydrogen i) Carbon: Burning of carbon to carbon dioxide (complete combustion)	
	 Consider formation of steam from ice at -100 C i) Solid phase- When the heat is added in ice which is at -100 C the temperature of ice increases to 00 C as shown in figure by process a-b.in this stage solid phase exists. ii) Solid+ Liquid phase- The point b is called is saturation point when heat is further added this heat cannot increase the temperature but ice is converted into water that means phase transformation takes place, thus in-between region b-c, solid and liquid phase exists. iii) Liquid phase- From point c-further heat is added up to 1000 C, in this region no phase change takes place, there is only liquid phase present. 	
b)	 provide customers with home improvements that enhance home comfort and increase property values for homeowners and businesses. Impact of energy conservation on Environment The other reason for conserving energy is the health and well-being of every life form on the entire planet. Using fossil fuels and some other energy forms typically pollute the environment in a number of ways. The air is polluted when fossil fuels like coal are burned and released into it Explain the process of formulation of steam from 0°C water with T-H diagram 	0.



	That means 1 kg of carbon needs 2.67 kg oxygen and produces 3.67 kg of carbon dioxide ii) Methane (CH4): $CH_4 + 2O_2 \qquad \longleftarrow CO_2 + 2H_2O$	
	i.e. $(12+1\times4)+2(16\times2)=(12+16\times2)+2(1\times2+16)$	00
	16 + 64 = 44 + 36	02
	$1 + 4 = \frac{11}{4} + \frac{9}{4}$	
	Burning of methane with oxygen to carbon dioxide and water/ steam i.e. That means 1 kg of methane needs 4 kg of oxygen to produce 11/4 kg of carbon dioxide and 9/4 kg of water /steam	
	iii)Hydrogen: $2H_2 + O_2 = 2H_2O$	01
	$2(1 \times 2) + (16 \times 2) = 2(1 \times 2 + 16)$	
	1 + 8 = 9	
	The union of hydrogen with oxygen produces steam it is represented by the following equitation 1 kg of hydrogen combines with 8 kg of oxygen to produce 9 kg of steam.	
d)	Explain higher and lower calorific value of a coal having following composition by mass Carbon 81%,Hydrogen 7%, Oxygen 8%,Nitrogen 2.5%, sulphur 1.5% and remaining is ash	04
	Carbon C = $81\% = 0.81$ Hydrogen = H2 = $7\% = 0.07$ Oxygen = O2 = $8\% = 0.08$ Nitrogen = N = $2.5\% = 0.025$ Sulphur = S = $1.5\% = 0.015$ Ash = $2.5\% = 0.025$	
	Dulong's formula: H.C.V. of coal = 33800 C + 144500 (H2 - O2/8) + 9300 S KJ / Kg =33800 x 0.81 + 144500 (0.07 - 0.08/8) + 9300 x 0.015 H.C.V. of coal = 36187.5 KJ / Kg	02

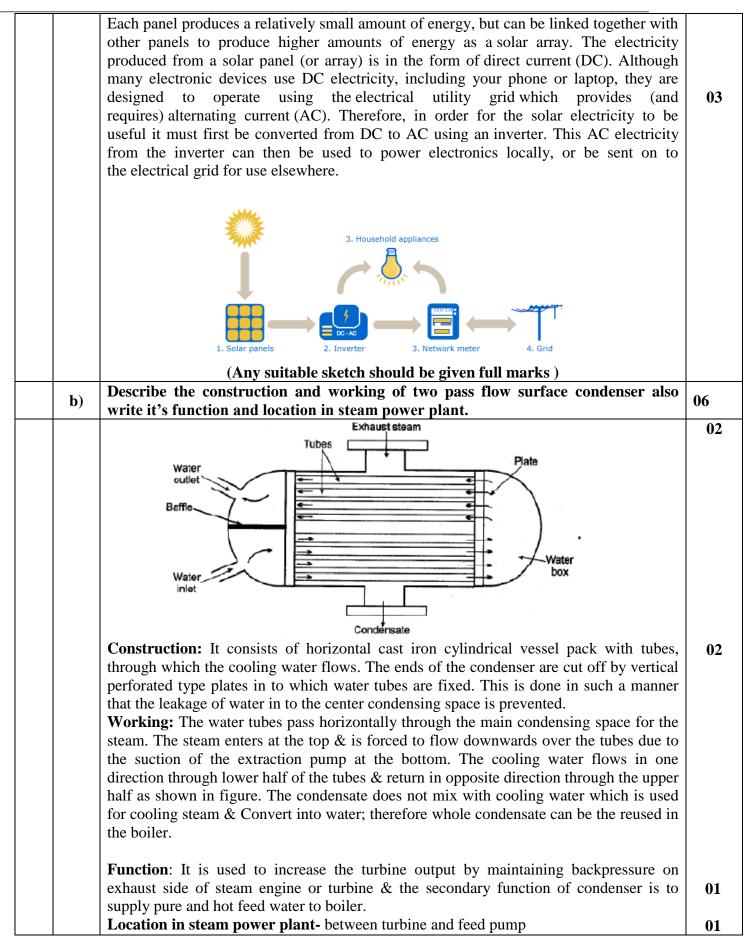


		L.C.V. of coal	0.2
		= H.C.V 9H2 x 2442 KJ / Kg 26187.5 0 = 0.07 = 2442	02
		$= 36187.5 - 9 \times 0.07 \times 2442$	
		L.C.V. of coal = $34649.04 \text{ KJ} / \text{Kg}$	
		Define	
		(i) LP (ii) D D	
	е		04
		(iii)Volumetric efficiency	
		(iv)Isothermal efficiency of air compressor	
			01
		i) I.D. It is the ratio of polytraphic work into speed of compressor in revolution per	01
		i) I.P. - It is the ratio of polytrophic work into speed of compressor in revolution per	
		second.	
		$I.P = W \times N$ watts	
		$\frac{1.1 - W \times 10}{60}$ watts	
		00	01
	Ans	ii) B.P. - It is the power required to drive the compressor or power delivered to the shaft	UI.
	Allo	of compressor	
		iii) Volumetric efficiency - It is the ratio of volume of free air delivery per stroke to the	
		swept Volume of piston.	
		iv) Compressor efficiency - For the same pressure ratio, It is the ratio of theoretical	01
		isothermal work to the actual work required to drive the compressor. OR It is the ratio of	• -
		isothermal power to the shaft power or brake power of motor required to drive the	
		compressor	
			01
5.		Attempt any TWO of the following	12
		(i) Draw the neat sketch of bomb calorimeter.	
		(ii) Write any three sources of air leakage in steam condenser.	
		(i)	
		Thermometer	
		← Copper	
		Oxygen valve	
		Release valve	
	a)	Bomb Water	
			0.2
		Fuse wire	03
		Cotton	
		Sealing water	
		Mains	
		Rheostat Leads to fuse	



		(ii)	
		Sources of air leakage in condenser:	03
		i) Air leak through joints and packing.	05
		ii) Air also comes in condenser with the steam.	
		iii) In jet condensers dissolved air in the cooling water enters the condenser.	
		in jet condensers dissorred an in the cooring water enters the condenser.	
	b)	Explain the necessity of multi-staging and inter cooling in case of two stage compressor with PV diagram	06
		Necessity of multi staging with intercooling in air compression: The large pressure ratio gives rise in high compression ratio and high discharged temperature which produce adverse effect on the efficiency and performance of the system. In such application efficiency decreases and works done and power increases. So to get better performance and saving in work and power multi staging with intercooling is necessary.	03
		p_3 delivery pressure H.P. cylinder $pV^n = constant$	
		pr = constante	0.2
		p ₂ intercooler pressure	03
		L.P. cylinder $pV^n = constant$	
		p ₁ intake pressure	
		→ V	
	c)	Write the strength and limitation of biomass power plant.	06
		Strength 1) Raw material used as cow dung is easily available in villages, rural area at free of	03
		cost.2) Easy to operate and having less maintenance.	
		3) No additional Co2 emission to environment.	
		4) Digested matter used as fertilizer	
		Limitation	
		1)Effectively implemented only where open space is available	
		2) For producing gas it takes more time.	03
		3) Initial investment is required for construction of well.	
6.		Attempt any TWO of the following.	12
	a)	Explain the construction and working of electricity generation through photovoltaic system.	
		(Any suitable Answer should be given full marks)	03
		Construction- A photovoltaic (PV) system is a system composed of one or more solar	
		panels combined with an inverter and other electrical and mechanical hardware that	
		use energy from the Sun to generate electricity. PV systems can vary greatly in size from	
		small rooftop or portable systems to massive utility-scale generation plants. Although	
		PV systems can operate by themselves as off-grid PV systems, this article focuses on	
		systems connected to the utility grid, or grid-tied PV systems.	
		Working-	
		The light from the Sun, made up of packets of energy called photons, falls onto a solar	
		panel and creates an electric current through a process called the photovoltaic effect.	







	c)	(i) Explain zeroth law of thermodynamic with neat sketch. (ii) Write the procedure for Energy saving in household appliances.	06
		Zeroth law of thermodynamics – If two thermodynamic systems are each in thermal	
		equilibrium with a third, then they are in thermal equilibrium with each other	
			03
	• \	System A System B	
	i)		
		1. 2.	
		System C	
		 A & C are in thermal equilibrium B & C are in thermal equilibrium 	
		then 3. A & B are also in thermal equilibrium with each	
		other	
		(Any suitable sketch should be given full marks)	
		(Any suitable procedure should be given full marks)	
		 Procedure for Energy saving in household appliances. Adjust your day-to-day behaviors. 	
		Replace your light bulbs	
		• Use smart power strips. (LED)	
		• Install a programmable or smart thermostat.	
	ii)	Install a programmable or smart thermostat.Purchase energy efficient appliances like fridge, Ac etc	
	ii)		
	ii)	• Purchase energy efficient appliances like fridge, Ac etc	03