11920 2 Hours / 50 Marks

Instructions:

- (1) All Questions are *compulsory*.
- (2) Illustrate your answers with neat sketches wherever necessary.
- (3) Figures to the right indicate full marks.
- (4) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

1. Attempt any NINE:

 $9 \times 2 = 18$

- (a) Define:
 - (i) Matrix
 - (ii) Slag
- (b) Write two ores of iron with chemical formula.
- (c) Name the products of blast furnace.
- (d) Define heat treatment of steel.
- (e) Define corrosion. Give its types.
- (f) Name the metal oxide film which is more protective. Explain.
- (g) Give reason galvanised containers are are not used for storing food staff.
- (h) Write two functions of Extenders.
- (i) Write any four characteristics of good fuel.
- (j) Write any two applications of Bio-diesel.
- (k) Define fuel. Give its types.
- (1) Define Lubricant. Give its types.

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2. Attempt any FOUR:

 $4 \times 4 = 16$

- (a) Write chemical reactions taking place in zone of reduction of blast furnace.
- (b) Explain:
 - (i) Normalising
 - (ii) Tempering
- (c) Write composition of steel. Explain its classification based on percentage of carbon.
- (d) Write any four advantages and disadvantages of gaseous fuel.
- (e) Explain proximate analysis of coal.
- (f) Give chemical composition of Bio-gas. Write its two uses.

3. Attempt any FOUR:

 $4 \times 4 = 16$

- (a) Explain oxygen absorption method with labelled diagram.
- (b) Explain metal cladding with labelled diagram.
- (c) Differentiate between galvanising and tinning.
- (d) Explain fluid-film lubrication with diagram.
- (e) Define following:
 - (i) Oiliness
 - (ii) Volatility
 - (iii) Cloud point
 - (iv) Pour point
- (f) Explain following:
 - (i) Saponification
 - (ii) Emulsification