# 11819 3 Hours / 100 Marks

| Seat No. |  |
|----------|--|
|----------|--|

Instructions:

- (1) All Questions are *compulsory*.
- (2) Answer each next main Question on a new page.
- (3) Illustrate your answers with neat sketches wherever necessary.
- (4) Figures to the right indicate full marks.
- (5) Assume suitable data, if necessary.
- (6) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

## 1. (A) Attempt any SIX of the following:

12

- (a) Write any two advantages of DCDA process.
- (b) Give the name of catalyst used in manufacturing of sulphuric acid by DCDA process.
- (c) State any four industrial uses of H<sub>2</sub>SO<sub>4</sub>.
- (d) Write physical and chemical properties of sulphuric acid.
- (e) Write the reactions involved in manufacturing of sulphuric acid.
- (f) Write any four industrial applications of nitric acid.
- (g) Define electrolysis.
- (h) Give any four types of cement.

[1 of 4] P.T.O.

17314 [2 of 4]

## (B) Attempt any TWO of the following:

- (a) Draw flowsheet of manufacturing of HCl.
- (b) State Linde and Claud's principles.
- (c) Give list of pollutants from cement industry. Write a name of device used to control pollution in cement industry.

#### 2. Attempt any TWO:

16

8

- (a) Describe manufacturing of urea with neat flow diagram.
- (b) Explain manufacturing process of phosphorous trichloride (PC $l_3$ ) and phosphorous pentachloride (PC $l_5$ ).
- (c) Explain with neat flow diagram manufacturing of soda ash.

#### 3. Attempt any FOUR:

16

- (a) Distinguish between single superphosphate and triple superphosphate with respect to chemical formula, application and properties.
- (b) State leaching and give application of it.
- (c) Give the process of conversion of yellow phosphorous to red phosphorous.
- (d) Give the reactions involved in manufacturing of HCl by soda & salt process.
- (e) Write formula for sodiumamalgum ? State its role in caustic soda manufacturing process.
- (f) Give four industrial applications of soda ash.

| 1731 | 4                 | [3 of 4] |
|------|-------------------|----------|
| 4.   | Attempt any FOUR: |          |

- (a) Distinguish between diaphragm cell and mercury cell process.
- (b) Explain manufacturing process of phosphoric acid by  $\rm H_2SO_4$  leaching process.
- (c) Name engineering problems involved in manufacturing of soda ash.
- (d) Write the reaction involved in manufacturing of water gas. State composition of water gas.
- (e) What precautions is to be taken while handling acetylene gas?
- (f) Draw neat flow diagram of producer gas manufacturing process.

# 5. Attempt any TWO:

16

16

- (a) Describe the manufacturing process of ammonia gas with neat flow diagram.
- (b) Distinguish between water gas and producer gas with respect to
  - (i) raw materials
  - (ii) composition of gases
  - (iii) application of gases
  - (iv) calorific value
- (c) Explain manufacturing process of urea with neat flow diagram.

P.T.O.

17314 [4 of 4]

# 6. Attempt any FOUR:

16

- (a) Give four uses each of dry ice and oxygen.
- (b) Define hardening and setting of cement.
- (c) Distinguish between dry and wet process of cement.
- (d) Give the manufacturing process of hydrogen gas by using natural gas.
- (e) Write the reactions involved in manufacturing of phosphoric acid by HCl leaching process.
- (f) Give the importance of mixed fertilizers.