

17539

11819

3 Hours / 100 Marks

Seat No.

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- Instructions :**
- (1) All Questions are *compulsory*.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answers with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Assume suitable data, if necessary.
 - (6) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

1. (A) Attempt any **THREE** of the following : **12**
- (a) Define pH. List the types of electrodes used for pH measurement.
 - (b) Compare single beam and double beam filter photometer by any four points.
 - (c) Name four blood gas parameters, with their normal range.
 - (d) List the types and concentration of various gas pollutants.
- (B) Attempt any **ONE** of the following : **6**
- (a) Draw block diagram of Analytical Instrument. Explain its each block.
 - (b) Explain with diagram basic parts of gas chromatography. Name two carrier gases used in gas chromatography.

2. Attempt any FOUR of the following :**16**

- (a) Draw the schematic diagram of multichannel photometer and explain it in brief.
- (b) Classify liquid chromatography in detail.
- (c) Explain measurement of nitrogen oxide using CO laser with a neat block diagram.
- (d) Explain spectrachemical shift in NMR.
- (e) Define the following :
 - (i) Environment
 - (ii) Air Pollution
- (f) Explain the working of thermal conductivity analyser using thermistor with diagram.

3. Attempt any FOUR of the following :**16**

- (a) State the applications of flame photometry. (any 4)
- (b) State the principle of NMR spectrometer. State any two applications of it.
- (c) Describe the construction of glass electrode with a neat diagram.
- (d) Explain why column temperature is so critical in gas chromatography.
- (e) Describe a technique for Ozone measurement.

4. (A) Attempt any THREE of the following :**12**

- (a) Draw a neat block diagram of liquid chromatography. What is the role of high pressure pump in it ?
- (b) State any four drawbacks of IR analyzer.
- (c) Draw a neat block diagram of a complete blood analyzer.
- (d) List two applications each of : (i) GCMS (ii) LCMS.

- (B) Attempt any ONE of the following :** **6**
- (a) Describe the conductivity method for measurement of SO_2 in air with a neat labelled diagram.
 - (b) Draw a neat labelled diagram of flame photometer. Explain the function of each block in brief.
- 5. Attempt any FOUR of the following :** **16**
- (a) Explain interaction of radiation with matter.
 - (b) Explain the working of a null detector type pH meter with a neat diagram.
 - (c) How do you convert volumetric concentration of gas to gravimetric concentration of gas ?
 - (d) Explain what is the effect of blood on electrode. State the use of buffer solution in brief.
 - (e) Explain any one technique of measurement of CO concentration in air.
 - (f) Explain principle of chromatography. Classify Gas Chromatography.
- 6. Attempt any FOUR of the following :** **16**
- (a) Draw a neat labelled diagram of GCMS.
 - (b) What is electrophoresis ? Explain paper electrophoresis in brief.
 - (c) Differentiate between colorimeter and flame photometer. (any four points)
 - (d) Explain discharge type atomizer used in flame photometer.
 - (e) Compare Gas Chromatography and Liquid Chromatography. (any four points)
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